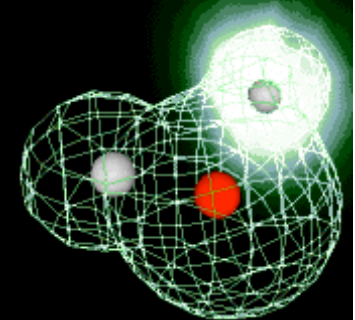
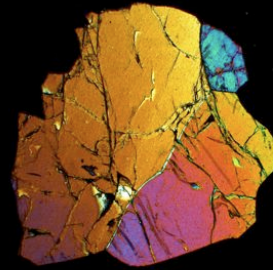
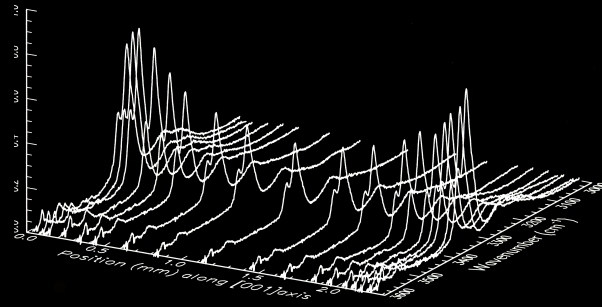


Water in the upper mantle: consequences to rheology

¹Hydrogen ¹
H
1.008

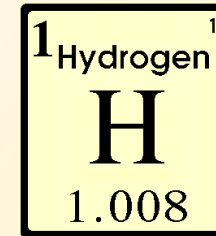


Sylvie Demouchy
CNRS, Geosciences Montpellier, France

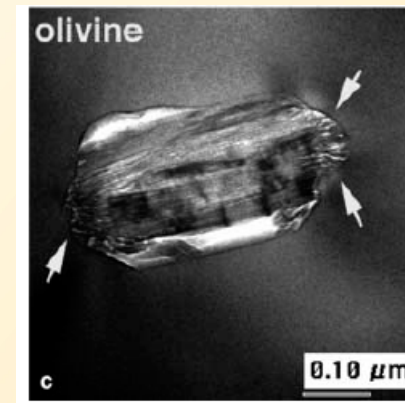
- *What is «water»* H_2O, OH^-, H^+, H^2
- *Where is this «water » in the mantle?*
- *Water & Deformation*

Where ????

- **Hydrogen as an impurity at the atomic scale in point defect:**
Hydrogen occupy a vacant site or is interstitial



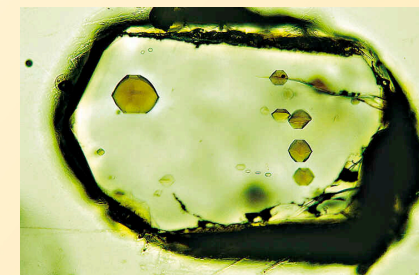
- **Microscopic scale (2 dimensions defect):**
OH or H₂O in hydrous phase
Amphiboles lamellae in pyroxenes,
Chlorite, humites or serpentines in olivine.



(Khisina et al, 2001)

- **Macroscopic scale (3 dimensions defect):**
H₂, OH, H₂O in fluid and/or glass inclusions.

Hydrogen is an incompatible element : it prefers to be in the magma than in the mineral !



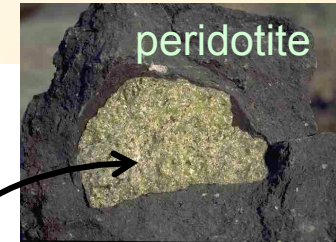
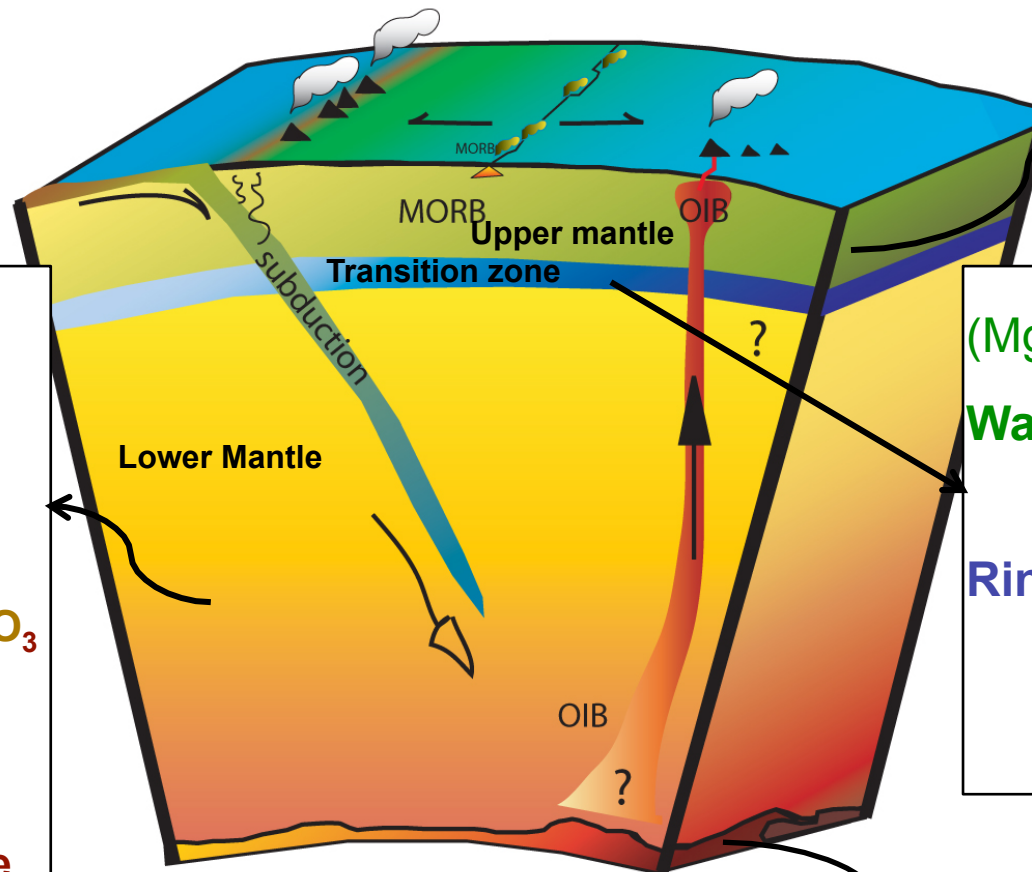
(Sobolev's web site)

- No molecular water in the minerals of the deep Earth
But, we will call it « water » anyway !
- “Water” solubility or “water” capacity is the maximum amount of hydrogen you can put into a mineral structure. It is determined experimentally. $f(T, P, \text{chemical composition}, fO_2, a_{\text{SiO}_2}, \text{co-existing phases})$
- The water solubility is NOT equal to the **REAL** hydrogen concentration in the mantle



Where?

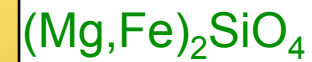
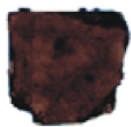
= Nominally Anhydrous Minerals



“Perovskite”



Ferropericlase



Wadsleyite



Ringwoodite



Majorite

Post-perovskite

Real water

MORB magma = 0.1 wt% H₂O

OIB magma = 0.3-1.0 wt% H₂O

Extrapolation

MORB source = 50-450 wt ppm H₂O

OIB source = 400-1000 wt ppm H₂O

UM xenoliths = 100-250
wt ppm H₂O

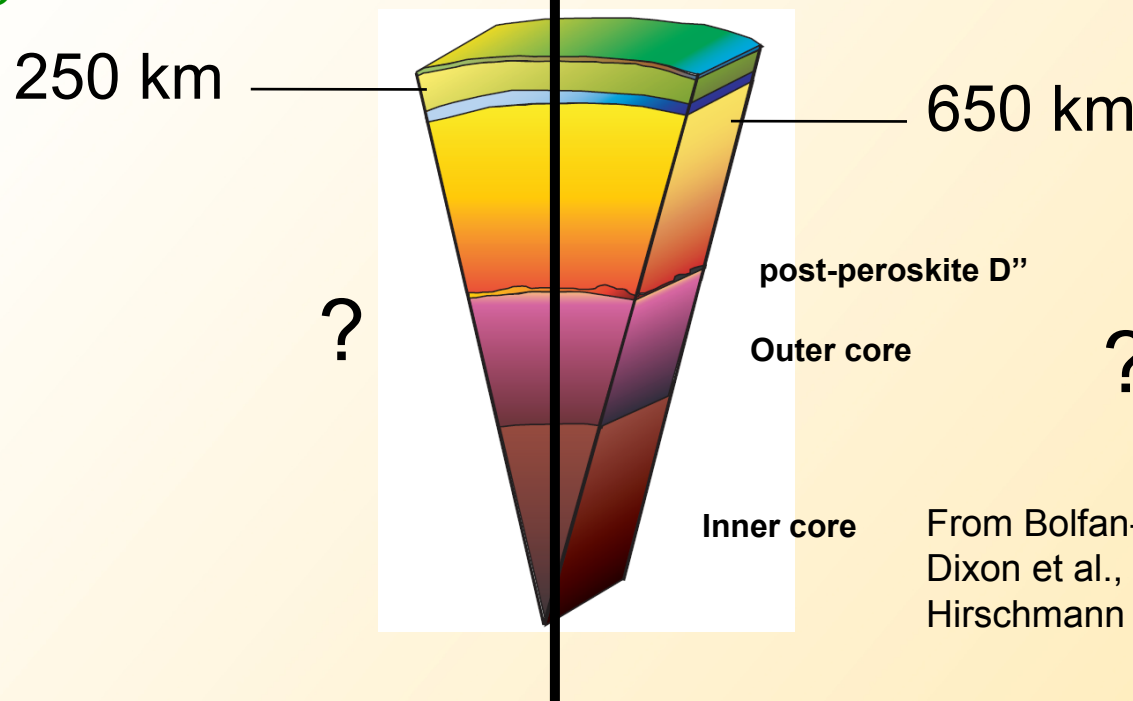
Water solubility

Fonction of T, P, chemistry, etc

UM = 250-800 wt ppm H₂O

TZ = 10 000 wt ppm H₂O = 1 wt%

LM = ~ 10 wt ppm H₂O



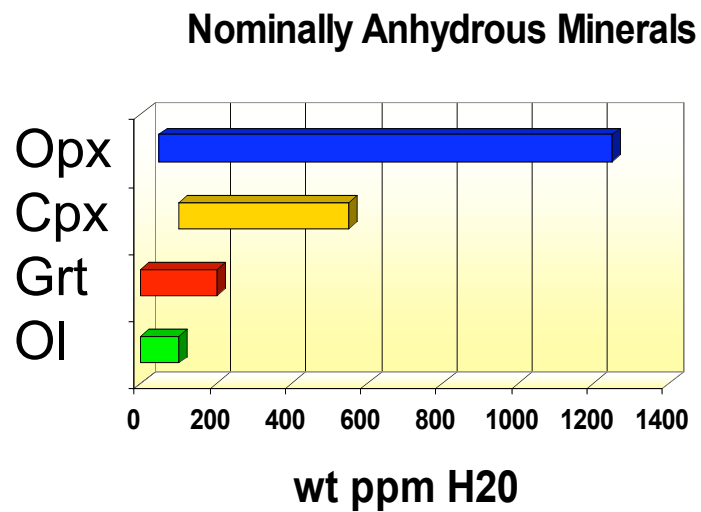
From Bolfan-Casanova, 2000;
Dixon et al., 2004 ;
Hirschmann et al, 2005

Nota: 1 wt% H₂O = 10 000 wt ppm H₂O
For olivine Fo₉₀, 1 wt ppm H₂O = 16.35 H/10⁶Si

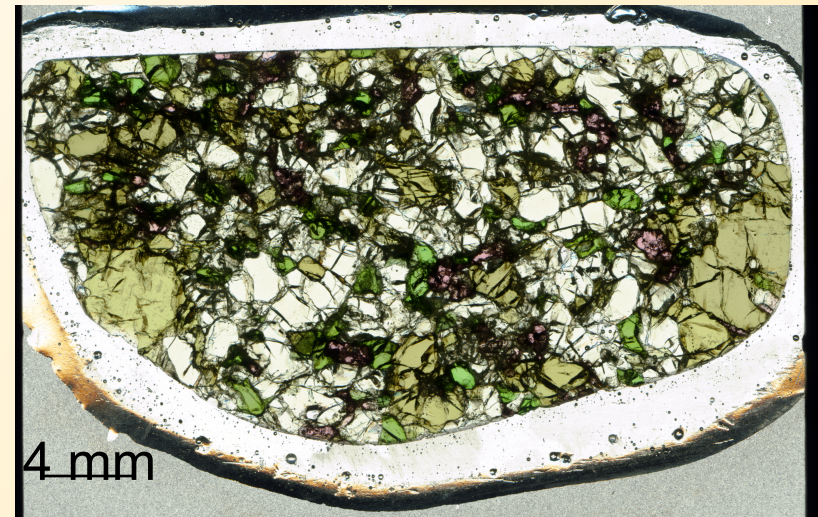
Close up

Water concentration of the upper mantle

- Between 0 wt ppm H₂O and 250 wt ppm H₂O

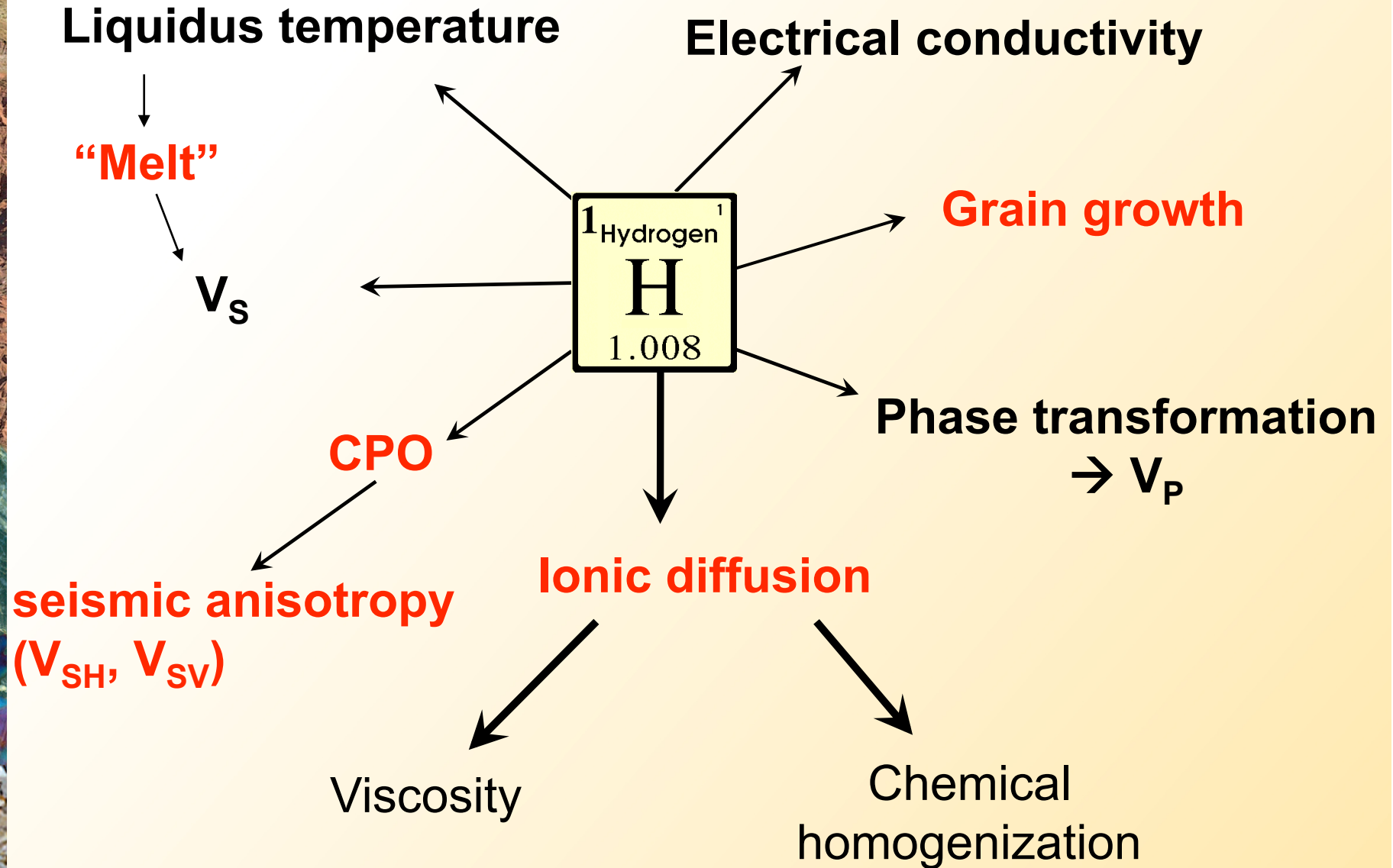


(Modified from Ingrin and Skogby, 2000)



Demouchy, 2004

EFFECT OF HYDROGEN



Water and deformation

To simplify, 2 deformation mechanisms
=> diffusion and dislocation

- *Effect of water in the diffusion creep regime*

Transport of matter by ion diffusion

- *Effect of water in the dislocation creep regime*

Transport of matter by dislocation motion

Different styles of motion: glide, climb and cross-slip....

We will see dislocation climb and glide

Power flow laws:

Empirical equations to describe deformation processes

$$\dot{\varepsilon} = A f_{H_2O}^r \frac{\sigma^n}{d^p} \exp\left(-\frac{E + PV}{RT}\right)$$

for diffusion creep $n = 1$

$$\dot{\varepsilon} = B f_{H_2O}^r \frac{\sigma^n}{d^p} \exp\left(-\frac{E + PV}{RT}\right)$$

for dislocation creep $n \approx 3$

Hydrogen uses point defect in the structure
Those point defect are moving → transport of matter

- From point defect (0 dimension defect in the lattice)

$$D_{\text{ion}} = C_{\text{vacancy}} \times D_{\text{vacancy}}$$

Effect of water in the diffusion creep regime

The presence of H increases the concentration of vacancy due to electroneutrality required by point defect theory to keep a crystal stable.

$$C_{\text{vacancy}}^{\text{hydrous conditions}} \neq C_{\text{vacancy}}^{\text{dry conditions}}$$

$$C_{\text{vacancy}}^{\text{hydrous conditions}} \gg C_{\text{vacancy}}^{\text{dry conditions}}$$

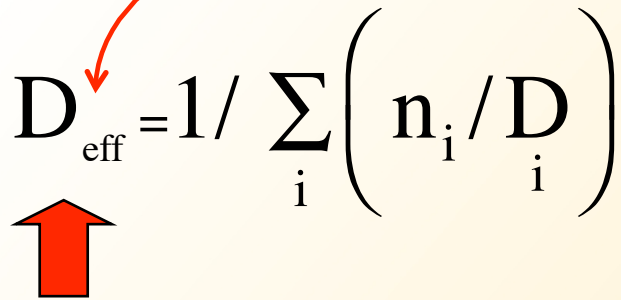
$$D_{\text{ion}} = C_{\text{vacancy}} \times D_{\text{vacancy}}$$

$$D_{\text{ion}}^{\text{hydrous conditions}} > D_{\text{ion}}^{\text{anhydrous conditions}}$$

Viscosity in the diffusion creep regime

• $\frac{1}{\eta} = A \frac{\Omega}{RT} \frac{D_{\text{eff}}}{G^2} \quad \text{or} \quad \eta = \sigma / \dot{\epsilon} = (\text{stress} / \text{strain rate})$

• $D_{\text{eff}} = 1 / \sum_i \left(n_i / D_i \right)$



Controlled by the slowest species in the mineral

To test and quantify this hypothesis

→ **diffusion** experiments and **deformation** experiments

Diffusion experiments

Si and O self-diffusion in olivine

2 to 4 orders of magnitude faster!

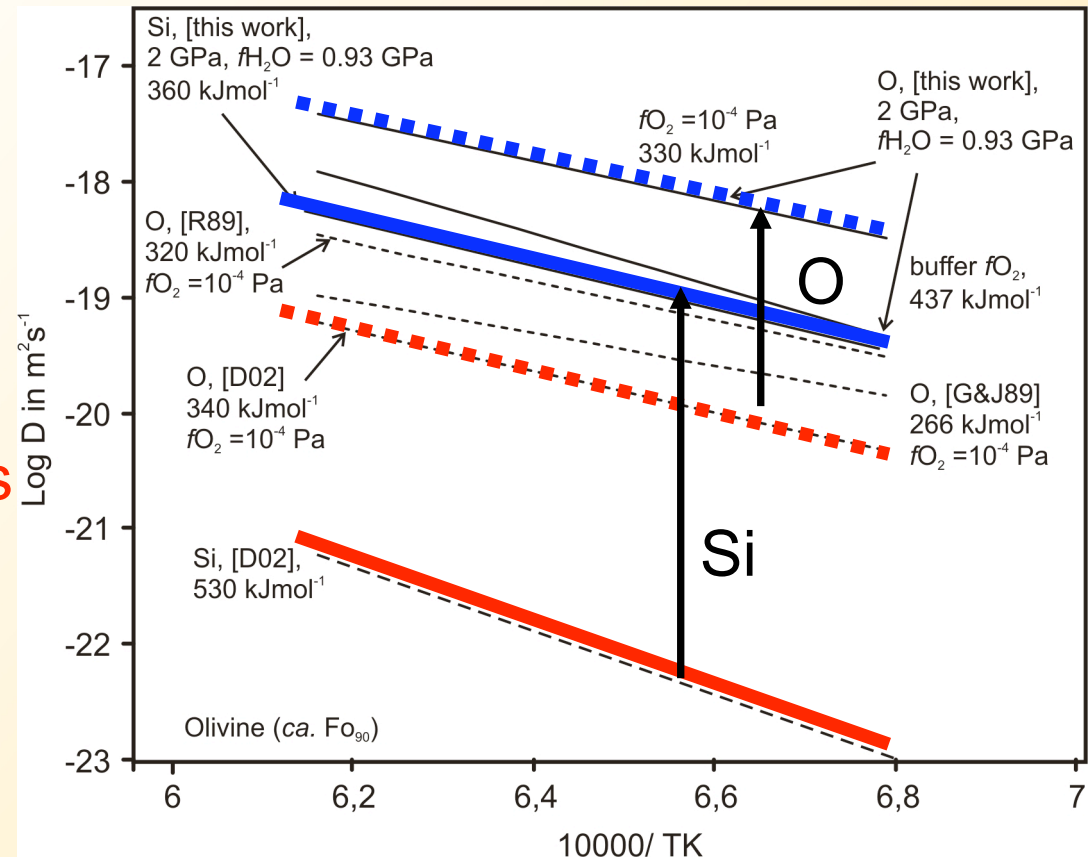
YES ! Diffusion is faster under hydrous conditions

Viscosity will be lower

Nabarro's model:

$$\eta = 9,8 \times 10^{21} \text{ à } 1,6 \times 10^{18} \text{ Pa.s}$$

@ 1200°C, 2GPa



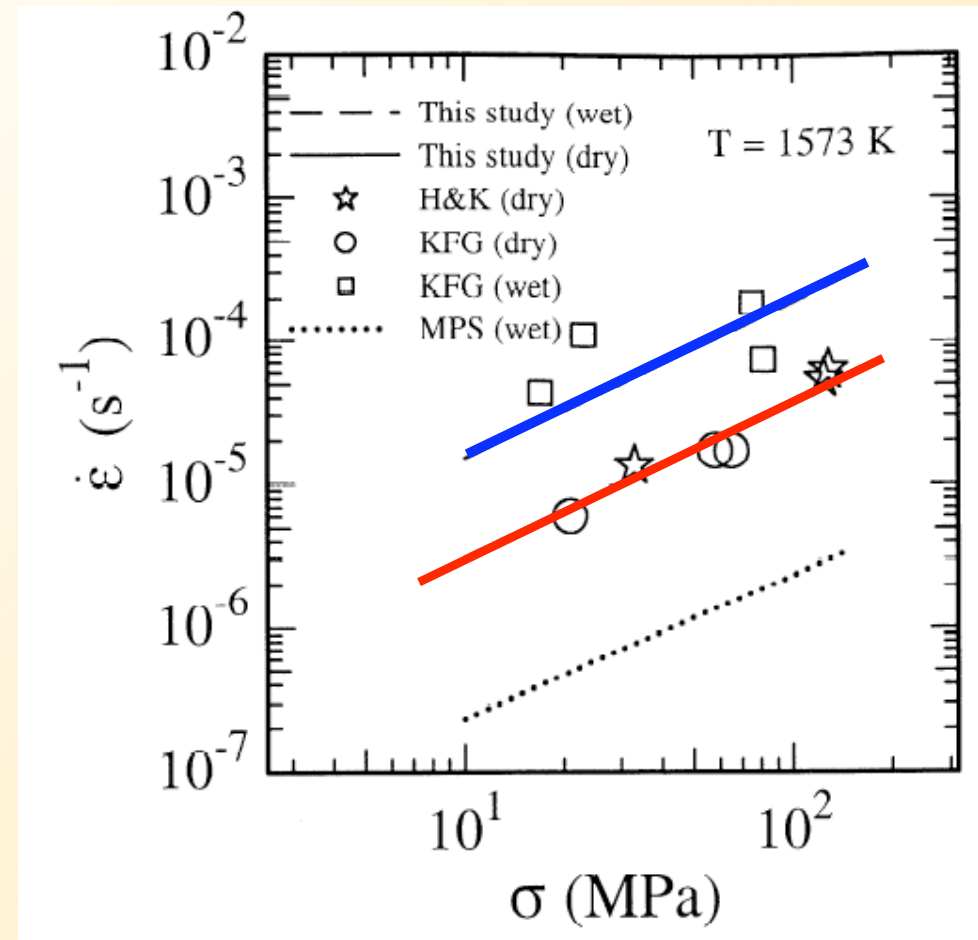
Costa and Chakraborty, 2008

Deformation experiments in the diffusion creep regime

- YES ! Stress is weaker under hydrous conditions

$$\eta = \sigma / \dot{\epsilon}$$

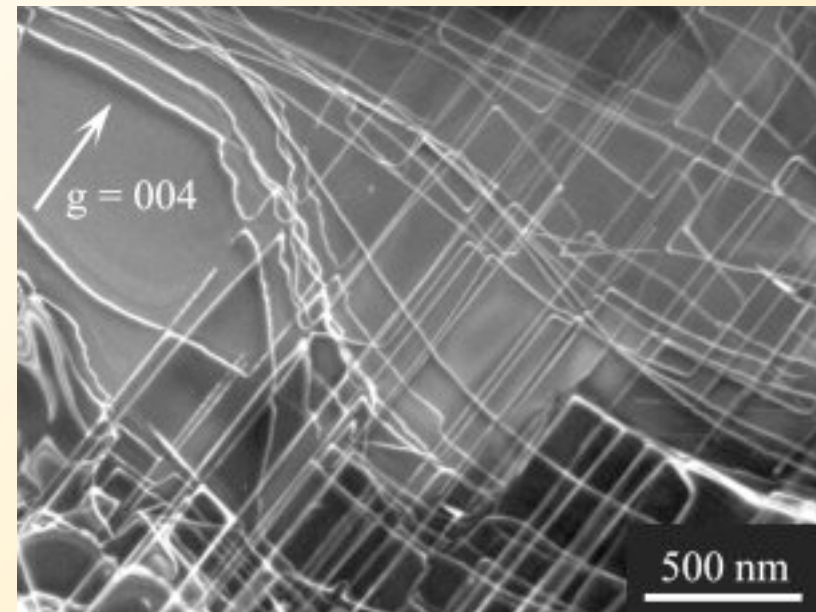
Viscosity will be lower !



Mei and Kohlstedt, 2000a

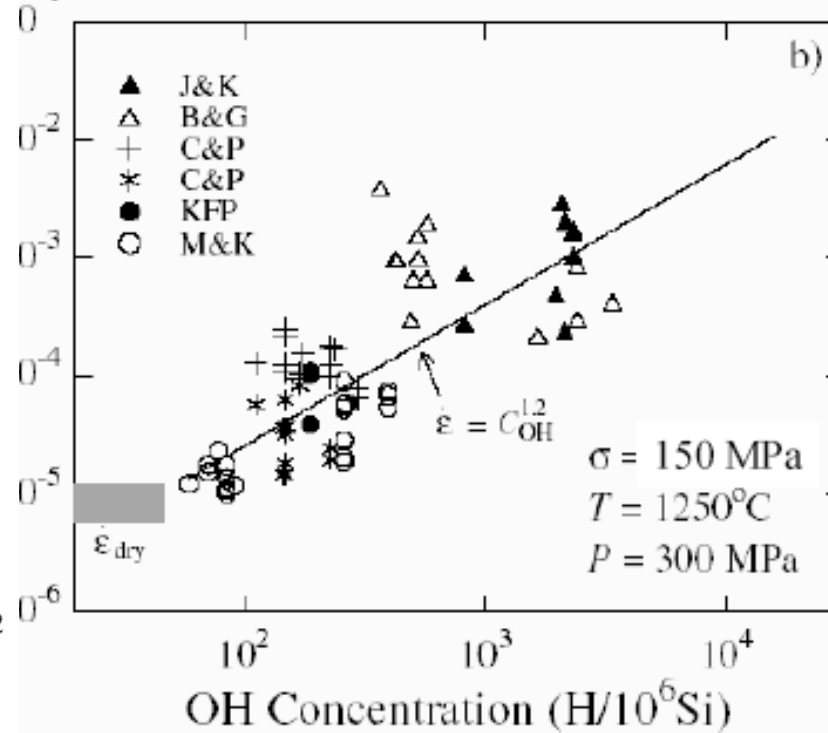
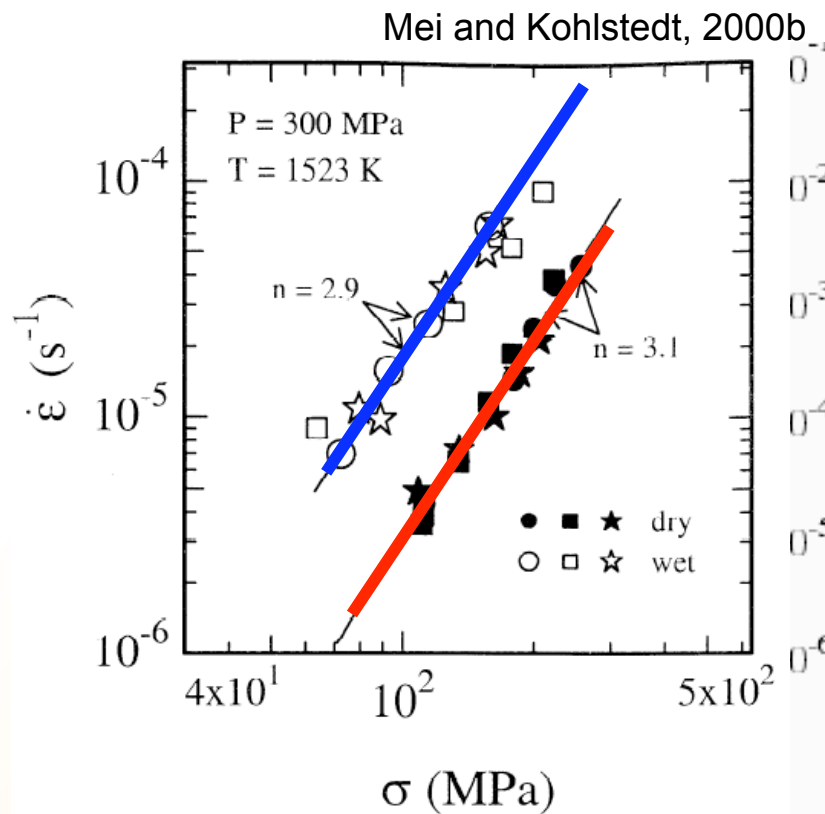
Effect of water in dislocation creep regime

- Transport through the lattice using dislocation
- Hydrogen as point defect
0 dimension defect
- Dislocations are
1 dimension defect



TEM image of deformed olivine

Deformation experiments

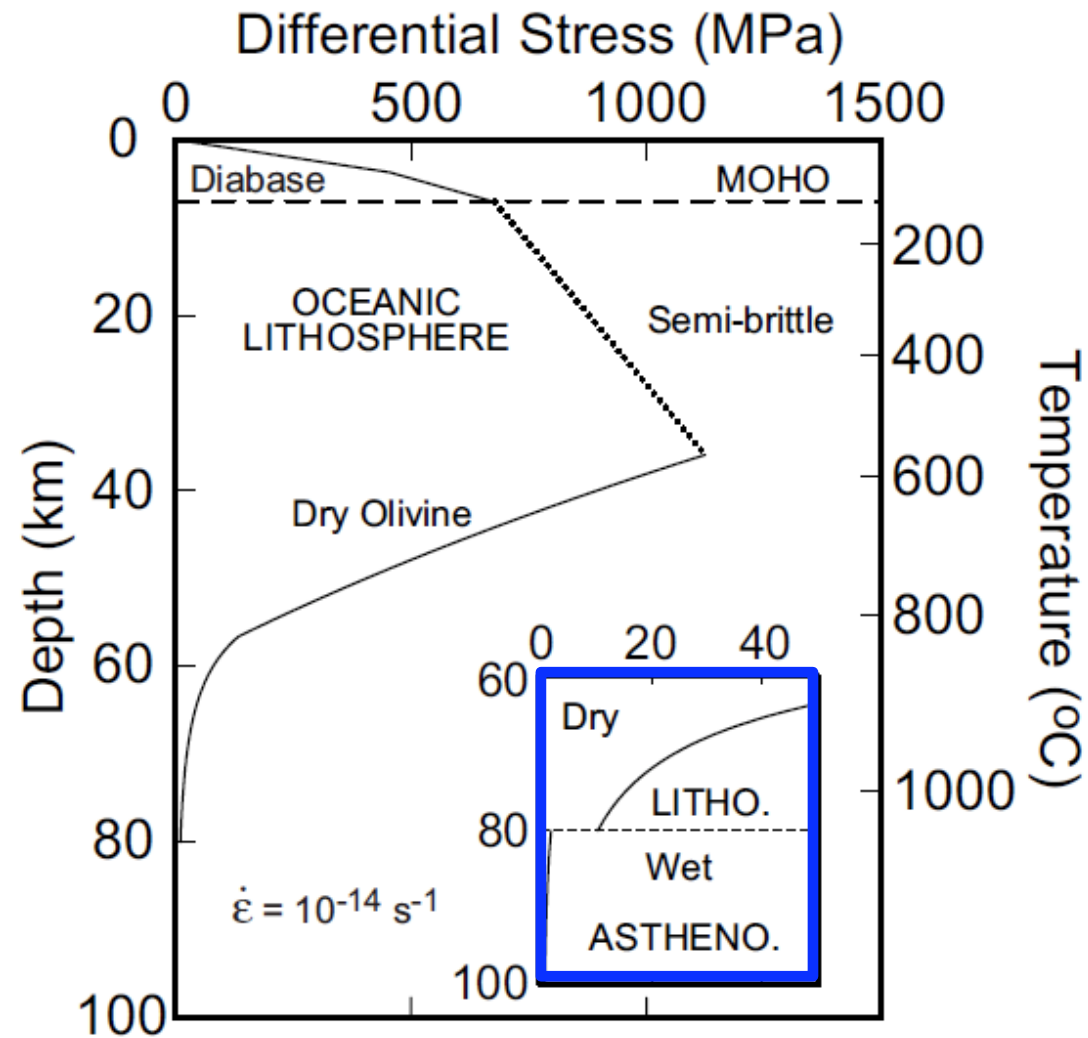


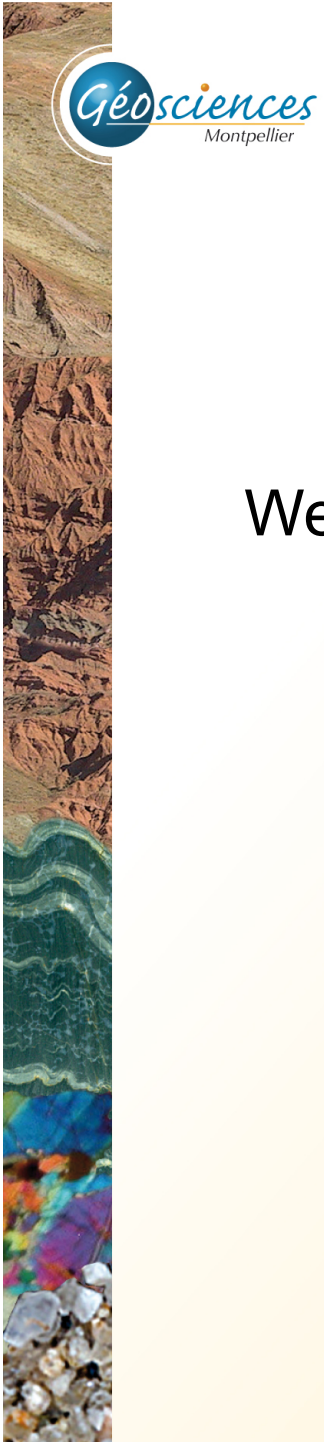
Hirth & Kohlstedt, 2003

Jung & Karato, 2001; Borch & Green, 1989; Chopra & Paterson, 1981;
Karato et al., 1986; Mei & Kohlstedt, 2000

**YES ! Stress is weaker
under hydrous conditions
Viscosity will be lower !**

Effect of water in dislocation creep regime



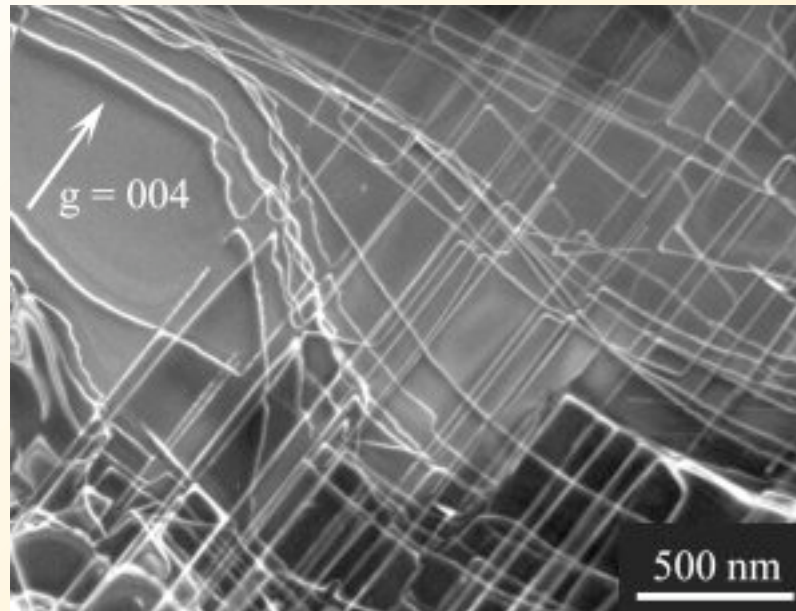


Effect of water in dislocation creep regime

Style of motion for dislocation:

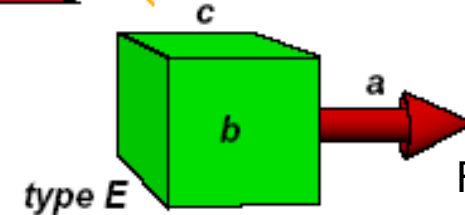
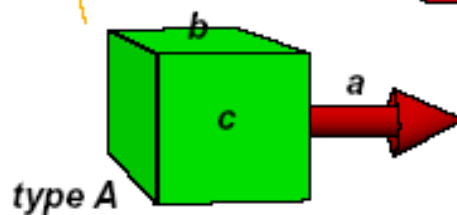
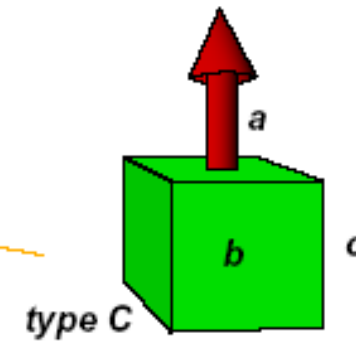
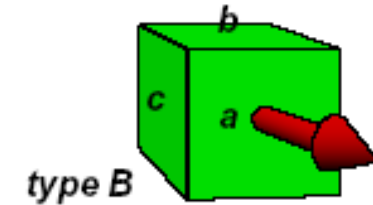
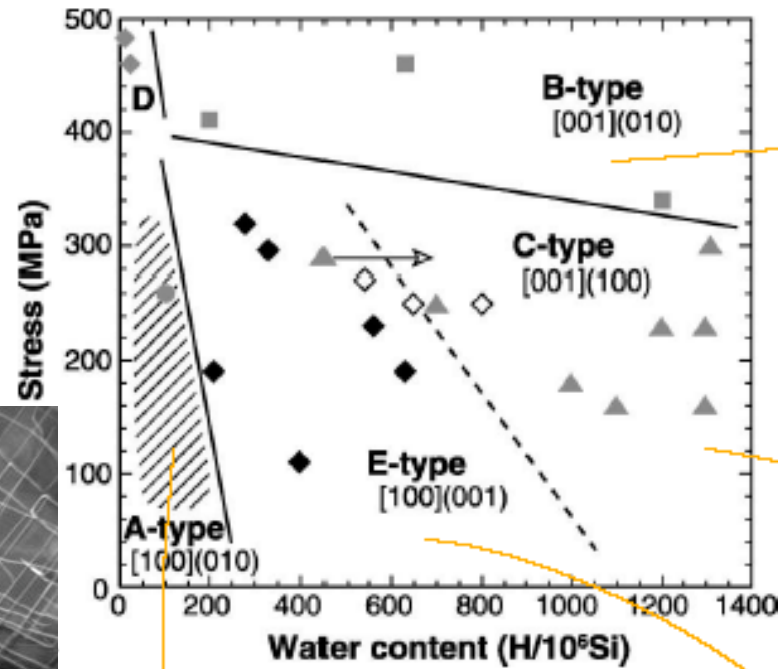
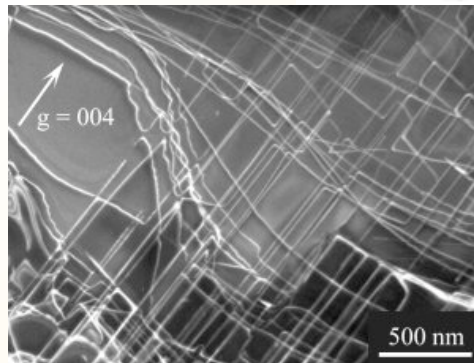
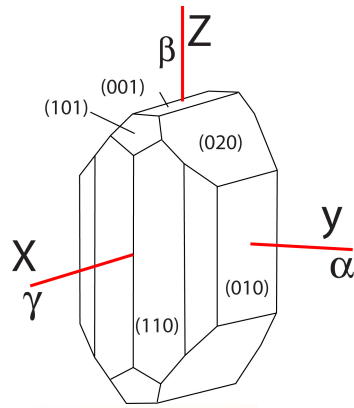
We saw the effect of water on dislocation climb

What's about dislocation glide?



Hydrogen in the lattice may affect the dislocation glide system

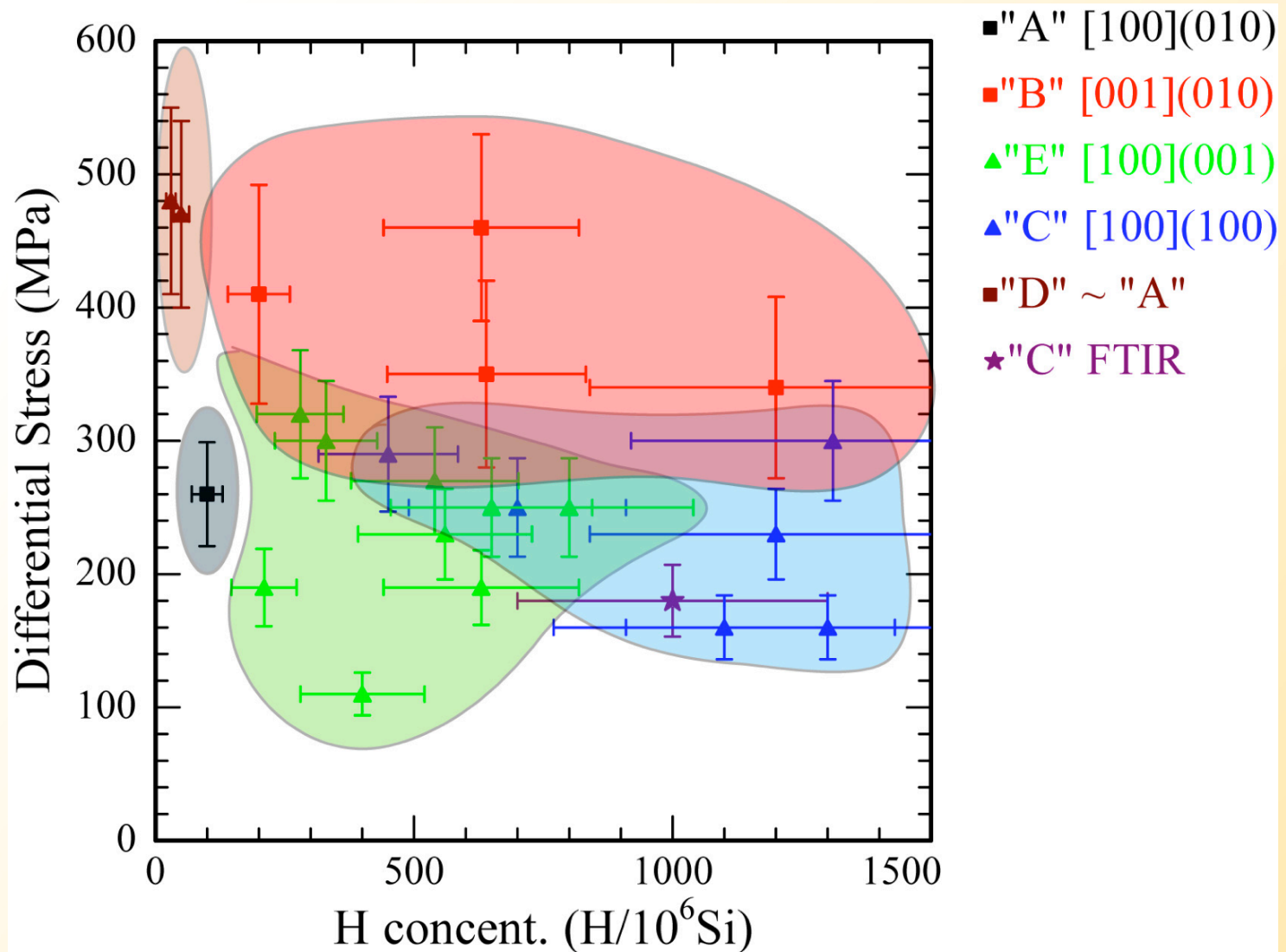
New type of olivine fabric from deformation experiments at modest water content and low stress
Katayama, Jung and Karato, Geology, 2004



From BKH

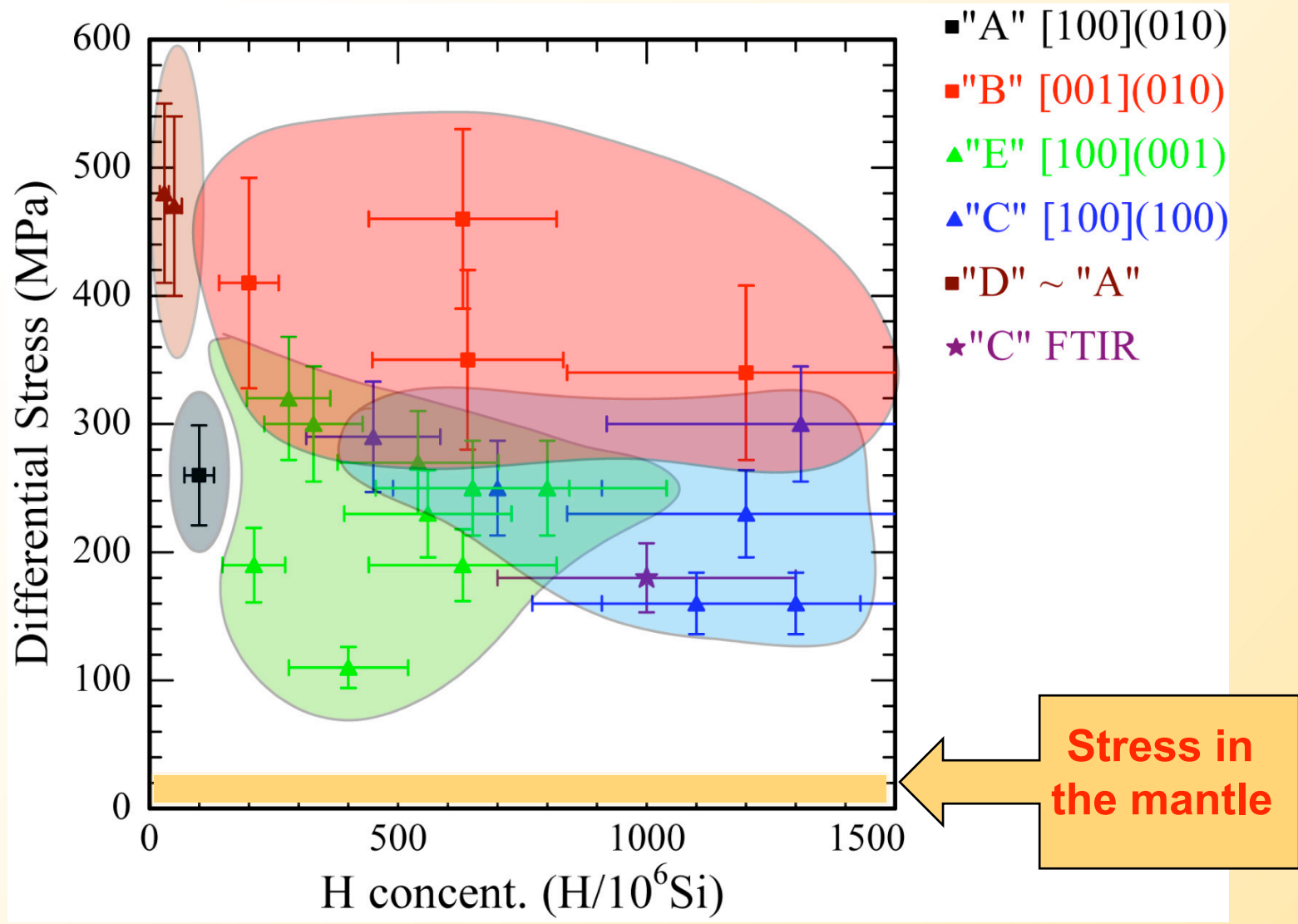
Effect of water in dislocation creep regime

With error bars.....



Effect of water in dislocation creep regime

With error bars.....



Conclusions

- Hydrogen is embedded in mantle minerals as impurities (and is fixed in the structure using **point defect**), **not molecular water !**
Deep Earth bulk hydrogen concentration....still poorly constraint !
- Hydrogen enhances ionic diffusion,
Then it weakens solids in the diffusion creep regime.
- Hydrogen affects dislocation velocity,
then it weakens solids in the dislocation creep regime.
- Hydrogen may change the dislocation glide system...

But keep your eyes open !